

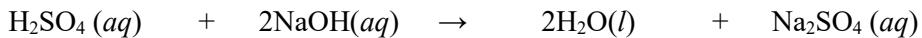
## C04 Double Replacement

### Double Replacement Reactions

#### Formula or Molecular Equation Version

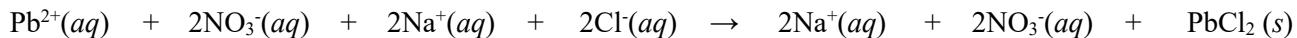


#### Formula or Molecular Equation

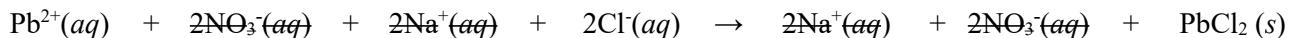


Return to this section after you have watched the Net Ionic Equation lesson.

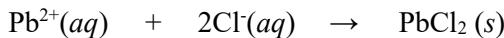
In aqueous solution the ionic compounds are dissociated.



The preferred method of writing this reaction is using **Net Ionic Equation** form, NIE. Where soluble ionic compounds are written in their dissociated form and the spectator ions are not written. There is a separate lesson on net ionic equations.



#### Net Ion Ionic Equation



#### Net Ionic Equation

The Net Ionic Equations for strong acids are not as simple as ionic compounds which just dissociate. We will cover acid net ionic equations later in the year.

